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### A New Polymorphism Variant: Nematic - Smectic C - Smectic A - Nematic

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A NEW POLYMORPHISM VARIANT: NEMATIC -  
SMECTIC C - SMECTIC A - NEMATIC

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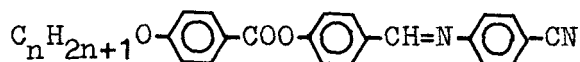
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Abstract: Some homologous 4[4-n-Alkyloxy-  
benzoyloxy]-benzylidene-4'-cyanoanilines are  
presented. Three of them exhibit a reentrant  
nematic phase. For the decyloxy compound the  
variant of polymorphism  $N_{re} S_C S_A N$  is was  
found.

As recently reported<sup>1</sup> 4[4-n-octyloxybenzoyloxy]-  
benzylidene-4'-cyanoaniline exhibits a reentrant ne-  
matic phase similar to the analogous stilbene<sup>2,3</sup>.  
Now we present some additional members of this homo-  
logous series. The substances were synthesized by  
condensation of 4-formyl-phenyl 4'-alkyloxy-benzoates<sup>4</sup>  
and 4-cyanoaniline in ethanolic solution. The sub-  
stances were purified by recrystallization from etha-  
nol or amylalcohol.  
The transition temperatures observed with a polariz-  
ing microscope are given in table 1.

TABLE 1



n	cr	N <sub>re</sub>		S <sub>C</sub>	S <sub>A</sub>	N	is.
4	.	137.5	-	-	.	(101)	. 299 .
5	.	122	-	-	.	(102)	. 280 .
6	.	115	-	-	.	( 84)	. 272 .
7	.	115	-	-	.	( 68)	. 265 .
8	.	108	.	152	-	.	198 . 255 .
9	.	97	(.	94)	-	.	224 . 247 .
10	.	100	(.	66	.	79)	. 232 . 242 .

cr : solid crystal      The numbers are the  
S<sub>A</sub>, S<sub>C</sub>: smectic A, C      transition temperatures  
N : nematic      (°C). Brackets denote  
is. : isotropic      monotropic phases.

As seen from the table, the first four homologues show nematic phases with a large temperature range and also metastable S<sub>A</sub> phases. In the octyloxy compound additionally a stable reentrant nematic phase (N<sub>re</sub>) occurs. In the following members of the homologous series<sup>9,10</sup> metastable reentrant nematic phases exist. Surprisingly, in the decyloxy derivative, between the reentrant nematic phase and the S<sub>A</sub> phase a smectic C phase appears. To our knowledge this is the first case that a smectic C liquid crystal compound possesses a strong longitudinal dipole moment.

The identification of the  $S_A$  and the reentrant nematic phase has been made by texture observation and by the study of miscibility with suitable reference substances<sup>1</sup>.

Unfortunately, the  $S_C$  phase and also the reentrant nematic phase of the decyloxy compound could not be identified by miscibility studies, because in the phase diagrams they always were separated by other phases ( $S_A$ ) from the corresponding phase regions of a reference substance.

The occurrence of the  $S_C$  phase is indicated by texture observation: This phase appears either as a schlieren texture exhibiting only points with four dark brushes or as a broken fan-shaped texture. On cooling the  $S_C$  schlieren texture, the reentrant nematic phase shows also a schlieren texture. On cooling the broken fan-shaped texture of  $S_C$ , the nematic phase forms a paramorphic fan-shaped texture. The same texture was also observed in the nematic phase of the nonyloxy derivative.

The transition  $S_C \rightarrow$  reentrant nematic could be detected by differential scanning calorimetry (see table 2).

As can be seen further from the table 2, the transition enthalpies  $N_{re}/S_A$  and  $S_A/N$  distinctly increase from the  $C_8$  to the  $C_{10}$  homologues.

The new polymorphism variant was confirmed by electrooptical investigations. The nematic high-temperature phase was oriented in a planar texture, in which the molecule director is aligned parallel to the substrate surfaces. On slowly cooling down, this director

TABLE 2

Transition Enthalpies ( $\text{J mol}^{-1}$ ) for the  $\text{C}_8$ ,  $\text{C}_9$  and  $\text{C}_{10}$  Homologues

n	melt.	$N_{\text{re}}/S_{\text{C}}$	$S_{\text{C}}/S_{\text{A}}$	$N_{\text{re}}/S_{\text{A}}$	$S_{\text{A}}/N$	N/is.	
8	39660	-	-	17.5	15	1396	<sup>1</sup>
9	8350	-	-	130	380	1600	
10	43000	47	50 <sup>x</sup>	-	800 <sup>x</sup>	1600 <sup>x</sup>	

<sup>x</sup>The transition peaks are not sharp and therefore the analysis is more difficult

orientation remains in the  $S_{\text{A}}$ ,  $S_{\text{C}}$  and  $N_{\text{re}}$  phase. When an electric field is applied perpendicularly to the director of the N or  $S_{\text{C}}$  phase, a dielectric reorientation (Fréedericksz transition) occurs. It was found that the threshold voltage of the reorientation in the reentrant nematic phase was a little higher (2.5 Volts) than that observed in the high-temperature nematic phase (2 Volts). The threshold voltage of the Fréedericksz effect in  $S_{\text{C}}$  was found to be about 4 Volts. This threshold voltage is unusually low compared with other  $S_{\text{C}}$  substances <sup>5,6,7</sup>. Obviously this is the result of the high positive dielectric anisotropy. Furthermore we could observe the typical optical pattern of  $S_{\text{C}}$  during the dielectric reorientation characterized by the occurrence of large homogenous regions with uniform interference colours <sup>5</sup>.

The new compounds presented here act as good electron acceptors and form with suitable liquid crystalline

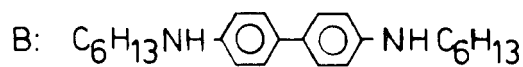
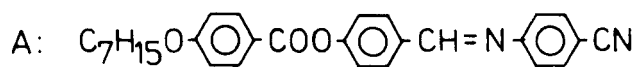
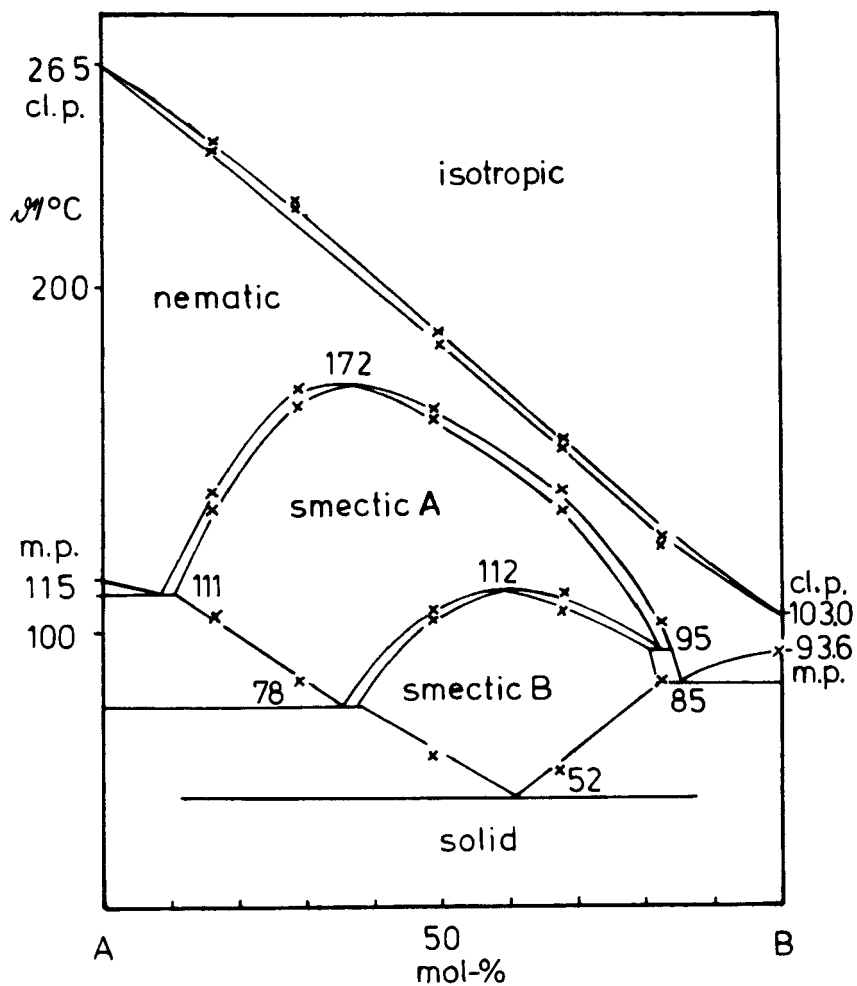


FIGURE 1

electron donators, e.g. 4,4'-bis-[n-alkylamino]-biphenyls,<sup>8</sup> stable electron-donator-acceptor-(EDA) complexes. Whereas the 4-[4-n-alkyloxy-benzoyloxy]-benzylidene-4'-cyanoanilines are slightly yellow and the 4,4'-bis-[n-alkylamino]-biphenyls are colourless, the mixtures of these compounds become strongly yellow due to the charge transfer band in the visible region. This effect is clearly seen in the boundary line of a contact preparation. Fig. 1 shows the isobaric diagram of state of the C<sub>7</sub> - compound and the 4,4'-bis-[n-hexylamino]-biphenyl<sup>9</sup> containing induced S<sub>B</sub> and S<sub>A</sub> phases with maxima in the transition curves. The different concentration positions of the two maxima indicate that the concentrations at the maxima must not always be identical with the composition of the complexes. Obviously by the EDA interaction, the longitudinal shifting of molecules is restricted and therefore the formation of smectic layers is favoured.

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